

## Note

### Alternative syntheses of 2,6-dideoxy-L-lyxo-hexose (2-deoxy-L-fucose) and its biochemical properties

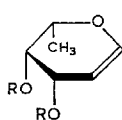
WALTER KORYTNYK, JANICE R. SUFRIN, AND RALPH J. BERNACKI

*Department of Experimental Therapeutics, Grace Cancer Drug Center, Buffalo, N.Y. 14263 (U.S.A.)*

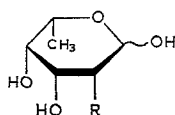
(Received August 19, 1981; accepted for publication, September 11, 1981)

There has been considerable interest in the synthesis of 2,6-dideoxy- $\alpha$ -L-lyxo-hexose (2-deoxy-L-fucose), because it is a constituent of several natural products<sup>1</sup>. Also, semi-synthetic anthracycline antibiotics with antitumour properties contain this sugar in place of the naturally occurring counterpart<sup>2</sup>. Our interest in this sugar stems from its close structural analogy with L-fucose, which occurs as a terminal sugar in many of the glycoconjugates of the cell membrane. 6-Deoxy-6-fluoro-L-galactose ("6F-L-fucose") was incorporated into the glycoconjugate<sup>3</sup>, and the formation of 6F-L-fucose 1-phosphate and GDP-6F-L-fucose has been indicated by h.p.l.c. of acid-soluble extracts of leukemic cells. 2-Deoxy-2-fluoro-D-fucose inhibited [<sup>14</sup>C]-L-fucose incorporation into mouse fibroblasts in culture<sup>4</sup>.

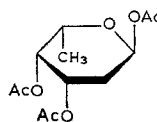
In considering methods for the synthesis of 2-deoxy-L-fucose, preference was given to procedures which would permit convenient introduction of tritium into a non-metabolisable position of the molecule for biochemical studies. 2-Deoxy-L-fucose was first synthesised<sup>5</sup> from the glycal **2** by acid-catalysed hydration and, more recently<sup>1,6</sup>, from ethyl 4,6-*O*-benzylidene-2-deoxy- $\alpha$ -D-ribo-hexopyranoside. However, neither method could be adapted for the introduction of tritium. Recently, improved methods for the synthesis of 2-deoxy sugars from glycals have been developed, using electrophilic additions to the double bonds; the most direct method<sup>7</sup> is the methanesulfonic acid-catalysed addition of acetic acid. Shortening the reaction time from that reported<sup>7</sup> improved the yield considerably and the conversion **1**→**4** could be effected in quantitative yield (t.l.c.). A similar result was obtained with 3,4,6-tri-*O*-acetyl-D-galactal (see Experimental).



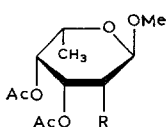
1 R = Ac  
2 R = H



3a R = H  
3c R = <sup>3</sup>H

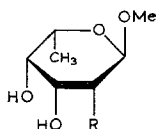


An alternative, less-direct synthesis of 2-deoxy sugars involves methoxymercuration of glycals, followed by borohydride reduction<sup>8,9</sup>. We have used this method for the synthesis of the title compound as well as for the introduction of deuterium and tritium at position 2. Methoxymercuration of **1** yielded solely methyl 2-acetoxymercuri-3,4-di-*O*-acetyl-2,6-dideoxy- $\alpha$ -L-talopyranoside (**5**). Methoxymercuration usually<sup>10</sup> proceeds by *trans* addition to the double bond, producing a methyl glycoside with the mercury component attached to C-2. Typically, mixtures of diaxial and diequatorial addition products are formed<sup>10</sup>. From the <sup>1</sup>H-n.m.r. spectrum of **5**, showing  $J_{1,2}$  and  $J_{2,3}$  values of 0.6 and 5.4 Hz, respectively, it can be concluded that H-2 is equatorial, and hence that MeO-1 and HgAc-2 are axial, and that only one product is formed. Also, the methoxyl protons in **5** resonate at a higher field ( $\delta$  3.34) than expected<sup>10</sup> for an equatorial group ( $\delta$  3.52). The <sup>199</sup>Hg-<sup>1</sup>H and <sup>199</sup>Hg-<sup>13</sup>C couplings have also been used for conformational assignments<sup>11</sup>. The  $J_{\text{Hg},3}$ ,  $J_{\text{Hg},2}$ , and  $J_{\text{Hg},1}$  values were 454, 188, and 90.5 Hz, respectively. Although these coupling constants were significantly larger than those found for methyl 3,4,6-tri-*O*-acetyl-2-chloromercuri-2-deoxy- $\alpha$ -D-talopyranoside (448, 176, and 86 Hz, respectively)<sup>11</sup>, the deviation is too small to be significant in terms of differences in the conformations of these two compounds [*i.e.*, <sup>1</sup>C<sub>4</sub>(L) and <sup>4</sup>C<sub>1</sub>(D), respectively].



**5** R = HgOAc

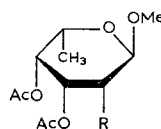
**6** R = H



**7a** R = H

**7b** R = <sup>2</sup>H

**7c** R = <sup>3</sup>H



**8a** R = H

**8b** R = <sup>2</sup>H

**8c** R = <sup>3</sup>H

The pseudo-axial AcO-4 in **1** exerts a marked influence on the stereochemistry of the electrophilic attack at C-2 and strongly favours L-*galacto* products. This applies to the acetic acid addition and the methoxymercuration reactions. When AcO-4 is pseudo-equatorial (as in 3,4,6-tri-*O*-acetyl-D-glucal), the approach of the reagents is less hindered, resulting in the loss of stereoselectivity.

Reduction of **5** with borohydride gave the acetylated methyl glycoside **8a**, which was deacetylated (**7a**) and then hydrolysed with acid, to give 2-deoxy-L-fucose (**3a**). An analogous reaction with borodeuteride gave **8b**, the configuration of which was established by n.m.r. spectroscopy. The  $J_{1,2}$  value of 1.5 Hz indicates the deuterium atom to be axial and that the deuteration had proceeded with retention of configuration. Also, in the 2-methylene region of the n.m.r. spectrum of **8a**, the proton resonating at higher field had been replaced with deuterium; for 2-deoxy sugars, H-2<sub>ax</sub> resonates at a higher field<sup>12</sup> than does H-2<sub>eq</sub>. The high degree of retention of configuration is noteworthy, in view of the free-radical nature of the reaction<sup>13</sup>. Borohydride reduction of methyl 2-acetoxymercuri-3,4,6-tri-*O*-acetyl-2-

TABLE I

EFFECTS OF 2-DEOXY-L-FUCOSE AND ITS DERIVATIVES ON CELL GROWTH AND MACROMOLECULAR BIOSYNTHESIS

Compounds	Leukemia L1210 (IC <sub>50</sub> , M)	Mouse mammary adenocarcinoma (TA3) (IC <sub>50</sub> , M)	P288 Leukemia		
			Growth (% control)	Incorporation (% control)	
				D-GlcN	L-Leu
3a	>10 <sup>-3a</sup>	>10 <sup>-3</sup>	96	110	107
4	—	>10 <sup>-3</sup>	78	81	81
7	>10 <sup>-3</sup>	>10 <sup>-3</sup>	96	110	107
6	5 × 10 <sup>-4</sup>	—	43	20	9
2	—	>10 <sup>-3</sup>	92	104	99
1	—	1.2 × 10 <sup>-4</sup>	53	43	27

<sup>a</sup>Growth was reduced by 30% at 10<sup>-3</sup>M.

deoxy-β-D-mannopyranoside was thought to proceed with retention of configuration, although the evidence was not convincing<sup>14</sup>. Reduction of **5** with borotritide gave **8c**, which was deprotected to give the tritiated 2-deoxy-L-fucose (**7c**).

At mM, 2-deoxy-L-fucose inhibited the growth of L1210 leukemia by 30%, had no effect on mouse-mammary adenocarcinoma (TA3) cells in culture (Table I), and inhibited the growth of SW613 mammary tumor cells to only a slight extent (11%). Since the last cell-line incorporated relatively large amounts of [<sup>3</sup>H]-labelled L-fucose, it was used for competition and incorporation studies. 2-Deoxy-L-fucose at mM was ineffective as a competitor of incorporation of L-fucose. The specific activity of the tritium label of **3c** was too low to allow assessment of its incorporation into the macromolecular fraction of SW613 cells.

The acetylated derivatives and precursors of 2-deoxy-L-fucose were active as growth inhibitors and exhibited a pronounced effect on the macromolecular biosynthesis of P288 leukemia cells in culture (Table I); the acetylated derivatives had the greater inhibitory activity (*cf.* **2** and **1**, **6** and **7**, and **4** and **3**). This situation has been observed with other carbohydrate analogues and is probably related to their enhanced lipid-permeability characteristics<sup>15</sup>. The effects of the analogues on the viability, the growth, and the incorporation of 2-amino-2-deoxy-D-glucose and L-leucine into the macromolecular fraction of P-288 cells were also determined. The growth-inhibitory compounds **6** and **1** also have a pronounced effect on the macromolecular biosynthesis, as seen by the decrease in incorporation of 2-amino-2-deoxy-D-glucose and L-leucine, but do not show selectivity with regard to the inhibition of precursor incorporation.

## EXPERIMENTAL

*General.* — Melting points were determined on a Mel-Temp apparatus and are uncorrected. I.r. spectra were recorded with a Perkin-Elmer 457 spectrophotometer and n.m.r. spectra for solutions in  $\text{CDCl}_3$  (internal  $\text{Me}_4\text{Si}$ ) with a Varian XL-100 spectrometer operating in the Fourier-transform mode. Optical rotations were measured with a Perkin-Elmer 141 polarimeter. T.l.c. was performed on silica gel with benzene-ether (1:1) and detection by charring with sulfuric acid. Solvents were removed by using a rotary evaporator under reduced pressure.

*1,3,4-Tri-O-acetyl-2,6-dideoxy-L-lyxo-hexopyranose (4).* — A solution of 3,4-di-O-acetyl-L-fucal (**1**; 1.60 g, 7.4 mmol) in acetic acid (15 mL) containing 67mm methanesulfonic acid was stirred at room temperature for 1 h, poured into chloroform, and neutralised with aqueous  $\text{NaHCO}_3$ . The organic layer was washed with water, dried ( $\text{CaCl}_2$ ), and concentrated, to give a colorless syrup (2.04 g) that gave only one spot in t.l.c. Short-path distillation (Kugelrohr,  $130^\circ/0.2$  Torr) gave **4** as an oil (1.80 g, 89%) which crystallised; m.p.  $198\text{--}200^\circ$  (from 2-propanol),  $[\alpha]_{\text{D}}^{22} -128.5^\circ$  (*c* 0.26, chloroform).

*Anal.* Calc. for  $\text{C}_{12}\text{H}_{18}\text{O}_7$ : C, 52.55; H, 6.51. Found: C, 52.73; H, 6.51.

*Methyl 2-acetoxymercuri-3,4-di-O-acetyl-2,6-dideoxy- $\alpha$ -L-talopyranoside (5).* — A solution of **1** (1.07 g, 5.0 mmol) and  $\text{Hg}(\text{OAc})_2$  (1.6 g, 6.0 mmol) in methanol (20 mL) was left at room temperature for 2 h. T.l.c. then showed that no **1** remained. After removal of the solvent, the residue was crystallised from 2-propanol, to give **5** (1.52 g, 60.3%), m.p.  $121\text{--}124^\circ$ ,  $[\alpha]_{\text{D}}^{22} -22^\circ$  (*c* 0.4, chloroform).

*Anal.* Calc. for  $\text{C}_{13}\text{H}_{20}\text{HgO}_8$ : C, 30.92; H, 3.99. Found: C, 31.13; H, 4.11.

*Methyl 3,4-di-O-acetyl-2,6-dideoxy- $\alpha$ -L-lyxo-hexopyranoside (6).* — A solution of **1** (1.45 g, 6.77 mmol) and  $\text{Hg}(\text{OAc})_2$  (2.16 g, 6.77 mmol) in methanol (20 mL) was left at room temperature in the dark for 3 h, cooled to  $0^\circ$ , and then treated with  $\text{NaBH}_4$  (350 mg) in small portions with stirring. After addition of some silica gel, the solvent was evaporated, and the residue was applied to a dry column of silicic acid. Elution with benzene-ether (1:1) afforded **6** (1.56 g, 93.4%) which, after crystallisation from ethanol-light petroleum, had m.p.  $66.5\text{--}67.5^\circ$ ,  $[\alpha]_{\text{D}}^{22} -166^\circ$  (*c* 0.79, chloroform).

*Anal.* Calc. for  $\text{C}_{11}\text{H}_{18}\text{O}_6$ : C, 53.65; H, 7.36. Found: C, 53.43; H, 7.29.

When  $\text{NaB}^2\text{H}_4$  was substituted for  $\text{NaBH}_4$ , **8b** was obtained in excellent yield.

*Methyl 2,6-dideoxy- $\alpha$ -L-lyxo-hexopyranoside (7a).* — To methanolic 1%  $\text{Ba}(\text{OH})_2$  (5 mL) was added **6** (530 mg, 2.1 mmol). The solution was kept at  $4^\circ$  for 17 h, neutralised with  $\text{CO}_2$ , filtered, and concentrated. Acetonitrile was evaporated from the residue, to give an oily product (310 mg, 89%) which was distilled [ $90^\circ$  (bath)/ $10^{-3}$  Torr] using a Kugelrohr. The oil crystallised on storage, and the hygroscopic crystals of **7a** had m.p.  $53\text{--}55^\circ$ ,  $[\alpha]_{\text{D}}^{22} -160^\circ$  (*c* 0.19, water).

*Anal.* Calc. for  $\text{C}_7\text{H}_{14}\text{O}_4$ : C, 51.83; H, 8.70. Found: C, 51.56; H, 8.73.

*2,6-Dideoxy-L-lyxo-hexose (3a).* — A solution of **7a** (1.48 g, 9.08 mmol) in 5mm  $\text{H}_2\text{SO}_4$  (25 mL) was heated for 6 h at  $70^\circ$ , cooled, neutralised with  $\text{BaCO}_3$ ,

centrifuged, and concentrated, to give an oily residue (1.32 g, 98%). Distillation ( $10^{-5}$  Torr) yielded a clear syrup, which eventually crystallised to give **3a**, m.p. 90–92°,  $[\alpha]_D^{22} -76^\circ$  (5 min)  $\rightarrow -57^\circ$  (3 h) (*c* 0.075, water); lit.<sup>5</sup> m.p. 103–106°; lit.<sup>10</sup> m.p. 92–94°,  $[\alpha]_D -75^\circ$  (5 min)  $\rightarrow -57^\circ$  (90 min) (*c* 1.5, water).

**2,6-Dideoxy-L-lyxo-[2-<sup>3</sup>H]hexose (3c).** — To a cooled, stirred solution of **5** (54 mg, 0.11 mmol) in ethanol (1 mL) was added sodium borotritide (0.9 mg, 0.02 mmol; 209 mCi/mg). After 30 min, the solvent was evaporated, and a solution of the residue in benzene–ether (1:1) was applied to a dry column of silica gel. Elution with benzene–ether (1:1, 35 mL) gave methyl 3,4-di-*O*-acetyl-2,6-dideoxy- $\alpha$ -L-lyxo-[2-<sup>3</sup>H]hexopyranoside (**8c**; 24 mg, 90.5%), which was dissolved in 95% ethanol (24 mL, 150 mCi/mmol). An aliquot (4 mL) was used for the biological studies and the remainder was concentrated. To the residue was added methanolic 1% Ba(OH)<sub>2</sub> (2 mL), and the solution was kept at 4° for 3 days. After the addition of 5M H<sub>2</sub>SO<sub>4</sub> (40 mL), the mixture was heated at 70° for 6 h, cooled, neutralised with BaCO<sub>3</sub>, filtered, and concentrated. A solution of the residue in ethanol was filtered and concentrated, to give **3c** (10 mg, 84%; 80 mCi/mmol) which co-chromatographed in t.l.c. with **3a**.

**1,3,4,6-Tetra-*O*-acetyl-2-deoxy- $\alpha$ -D-lyxo-hexose.** — 3,4,6-Tri-*O*-acetyl-D-galactal (1.5 g, 5.5 mmol) was treated with acetic acid (15 mL) containing 0.067M methanesulfonic acid, as described above for **1**. Work-up gave white crystals (1.2 g, 66%) that appeared to be homogeneous (n.m.r. spectrum). Recrystallisation from 2-propanol gave the title compound, m.p. 99–102°,  $[\alpha]_D^{22} +118^\circ$  (*c* 1.2, chloroform); lit.<sup>7</sup> m.p. 102–103°,  $[\alpha]_D +123^\circ$ ; lit.<sup>17</sup> m.p. 97°,  $[\alpha]_D +118^\circ$ .

**Biological evaluation.** — L1210 leukemia cells were grown in stationary tube cultures in RPMI 1640 medium<sup>18</sup> containing 10% of heat-inactivated, foetal calf serum. Murine P288 leukemia cells were maintained as an ascites tumor in DBA/2J female mice. Periodically, cells were removed from mice, under aseptic conditions, washed twice in RPMI 1640 medium, and cultured in RPMI 1640 containing 10% of foetal calf serum. These cultures were grown in stationary tube cultures in a 90% air/10% CO<sub>2</sub> incubator. They were maintained *in vitro* for 1–3 months, whereupon new *in vitro* cultures were initiated from the mouse ascites tumors.

(a) **L1210 system.** An inoculum of  $5 \times 10^4$  cells in 1 mL of RPMI 1640 medium (containing 10% of heat-inactivated, foetal calf serum and 20mM Hepes buffer) was supplemented with 1 mL of the same medium containing the compound to be tested. The tubes were incubated in an upright position for 3 days, and growth was estimated either by protein assay or cell counts (using a Coulter counter). The growth in control cultures varied from 6–10 fold after 3 days. Each concentration was tested in triplicate. For compounds found to be inhibitory, the tests were repeated at least twice. Variation between different tests was within  $\pm 10\%$  for the 50% inhibitory concentration. The results are expressed in terms of IC<sub>50</sub> (the molar concentration of the sugar analogue in the nutrient medium leading to 50% inhibition of cell growth).

(b) **P288 system.** For routine testing, P288 murine leukemic cells were suspended at  $\sim 10^5$  cells/mL in fresh RPMI 1640 (without D-glucose and containing 10% of

heat-inactivated, foetal calf serum). Aliquots (1 mL) were transferred to disposable polyethylene tubes and placed in a CO<sub>2</sub>/air incubator. After 1 h, sugar analogues were added to a final concentration of mM (or as otherwise stated). Cell growth and viability were monitored at later times by using a Coulter counter and Trypan blue dye exclusion, respectively. 2-Amino-2-deoxy-D-[<sup>14</sup>C]glucose (2 μM, 1.1 × 10<sup>6</sup> d.p.m.) and [<sup>3</sup>H]leucine (0.3 mM, 2.6 × 10<sup>6</sup> d.p.m.) were added to cell cultures to assess the effects of sugar analogues on protein and glycoprotein biosynthesis. Incubations were terminated 5 h later by the addition of 10% trichloroacetic acid (2 mL), and the resulting pellet was dissolved in NaOH and its radioactivity assayed by scintillation counting.

#### ACKNOWLEDGMENTS

This study was supported by USPHS Grants CA-08793, CA-19814 and CA-13038. We thank Ms. Onda Simmons for determining the n.m.r. spectra, and Ms. Patricia McKernan for skilful technical assistance in the cell-culture studies.

#### REFERENCES

- 1 D. HORTON, T. M. CHEUNG, AND W. WECKERLE, *Methods Carbohydr. Chem.*, 8 (1980) 201–205.
- 2 E.-F. FUCHS, D. HORTON, AND W. WECKERLE, *Carbohydr. Res.*, 57 (1977) c36–c39; E. F. FUCHS, D. HORTON, W. WECKERLE, AND E. WINTER-MIHALLY, *J. Med. Chem.*, 22 (1979) 406–411; H. S. EL KHADEM AND D. L. SWARTZ, *Carbohydr. Res.*, 65 (1978) c1–c2.
- 3 J. R. SUFRIN, R. J. BERNACKI, M. J. MORIN, AND W. KORYTNYK, *J. Med. Chem.*, 23 (1980) 143–149.
- 4 D. J. WINTERBOURNE, C. G. BUTCHARD, AND P. W. KENT, *Biochem. Biophys. Res. Commun.*, 87 (1979) 989–992.
- 5 B. ISELIN AND T. REICHSTEIN, *Helv. Chim. Acta*, 27 (1944) 1200–1203.
- 6 T. M. CHEUNG, D. HORTON, AND W. WECKERLE, *Carbohydr. Res.*, 58 (1977) 139–151.
- 7 D. M. CIMENT AND R. J. FERRIER, *J. Chem. Soc., C*, (1966) 441–445.
- 8 G. INGLIS, J. SCHWARTZ, AND L. McLAREN, *J. Chem. Soc.*, (1962) 1014–1019.
- 9 P. MANOLOPOULOS, M. MEDNICK, AND N. LICHTIN, *J. Am. Chem. Soc.*, 84 (1962) 2203–2210.
- 10 K. TAKIURA AND S. HONDA, *Carbohydr. Res.*, 21 (1972) 379–391.
- 11 V. G. GIBB AND L. D. HALL, *Carbohydr. Res.*, 55 (1977) 239–252.
- 12 M. MIYAMOTO, Y. KAWAMATSU, M. SHINOHARA, K. NAKANISHI, Y. NAKADAIRA, AND N. S. BHACCA, *Tetrahedron Lett.*, (1964) 2371–2377.
- 13 G. M. WHITESIDES AND J. S. FILIPPO, JR., *J. Am. Chem. Soc.*, 92 (1970) 6611–6624; C. L. HILL AND G. M. WHITESIDES, *ibid.*, 96 (1974) 870–876.
- 14 J. H. LEFTIN AND N. N. LICHTIN, *Isr. J. Chem.*, 3 (1965) 107–111; *Chem. Abstr.*, 64 (1966) 3667c.
- 15 R. J. BERNACKI, M. SHARMA, N. K. PORTER, Y. RUSTUM, B. PAUL, AND W. KORYTNYK, *J. Supramol. Struct.*, 7 (1977) 235–250.
- 16 S. TAKAHASHI, M. KURABAYASHI, AND E. OHKI, *Chem. Pharm. Bull.*, 15 (1967) 1657–1661.
- 17 A. A. CHALMERS AND R. H. HALL, *J. Chem. Soc., Perkin Trans. 2*, (1974) 728–732.
- 18 G. E. MOORE, R. GERNER, AND H. FRANKLIN, *J. Am. Med. Assoc.*, 199 (1967) 519–524.